SHRI RAMSWAROOD MEMORIAL UNIVERSITY

End Semester Examination (2021-22)-Odd Semester

| M.Sc. (Chemistry) I Year (I Sem.) | | | | | | | | | | | | |
|--------------------------------------|-----|--|---|--|---|----|------|-------|---------------|------|------|-------|
| Course Name: Analytical Chemistry Co | | | | | | | | Code | Code: MCY1005 | | | |
| Time: 02 Hours | | | | | | | | | Max | к Ма | rks: | 60 |
| University Roll No. | | | | | | | | | | | | |
| | 1 1 | | 1 | | 1 | (T | o be | fille | d by | the | Stuc | lent) |

Note: Please read instructions carefully:

- a) The question paper has 03 sections and it is compulsory to attempt all sections.
- b) All questions of Section A are compulsory; questions in Section B and C contain choice.

| Section A. Vorry Short Angeron terms Organisms | | | | | | |
|--|---|------|------|---------------|--|--|
| Section A: Very Short Answer type Questions Attempt all the questions. | | BL | CLO | Marks (10) | | |
| 1. | Distinguish between accuracy and precision. | BL 4 | CLO1 | 02 | | |
| 2. | What is determinate and indeterminate error? | BL 1 | CLO2 | 02 | | |
| 3. | What are the criteria's for Good Laboratory Practices? | BL 1 | CLO2 | 02 | | |
| 4. | Discuss the basic principles of chromatography. | BL 2 | CLO3 | 02 | | |
| 5. | Describe the applications of GC-MS. | BL 2 | CLO4 | 02 | | |
| Section B: Short Answer Type Questions | | | CLO | Marks | | |
| Atte | mpt any 03 out of 06 questions. | | | (30) | | |
| 1. | Discuss the significance of pH in analysis. | BL 2 | CLO1 | 10 | | |
| 2. | What are the buffers and their importance in chemical analysis? | BL 1 | CLO2 | 10 | | |
| 3. | How will you reduce errors? Discuss the benefits of quality system. | BL 1 | CLO4 | 10 | | |
| 4. | What are the applications of column chromatography in research? | BL 1 | CLO3 | 10 | | |
| 5. | Distinguish between quality audit and quality review. | BL 4 | CLO1 | 10 | | |
| 6. | What do you mean by instrumentation validation? Why it is necessary? | | CLO4 | 10 | | |
| Section C: Long Answer Type Questions/Case Study Attempt any 01 out of 03 questions. | | | CLO | Marks (20) | | |
| 1. | Explain the role of total quality management in analytical chemistry. | BL 2 | CLO1 | 20 | | |
| 2. | What is the solvent treatment system in HPLC? Explain isocratic and gradient systems. | BL 1 | CLO4 | 20 | | |
| 3. | What are the detectors? Explain the operative principles of coulometric, thermionic and NP detectors. | BL 1 | CLO3 | 20 | | |
